

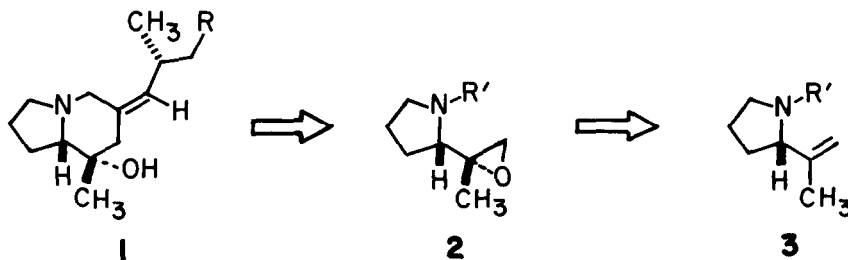
CARBONYL PARTICIPATION IN THE REACTION OF N-ACYL-2-ISOPROPENYLPYRROLIDINES  
WITH HALOGEN ELECTROPHILES. STEREOCONTROLLED PREPARATION OF ENANTIOMERICALLY  
PURE SYNTHONS FOR THE SYNTHESIS OF PUMILIOTOXIN A ALKALOIDS.

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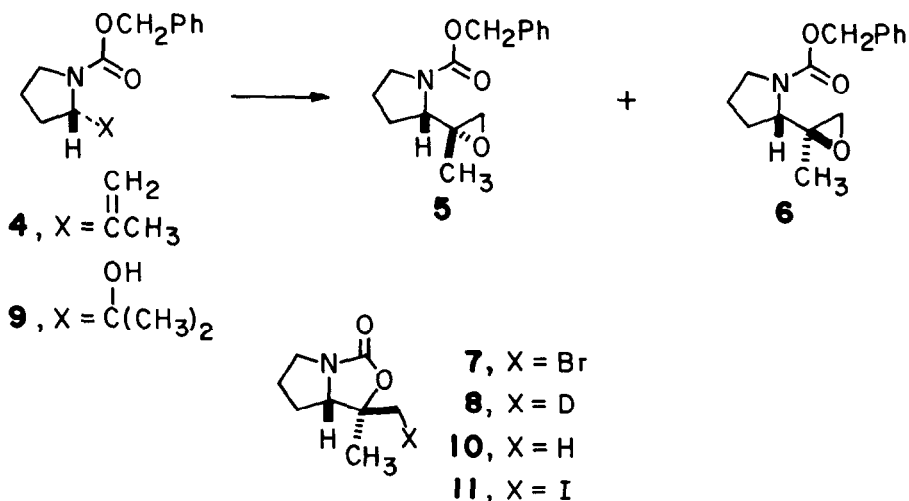
**Summary:** Allylic carbamate 4 and allylic benzamide 12 react with halogen electrophiles with high 1,2-relative asymmetric induction via carbonyl participation.

The control of acyclic 1,2-stereochemical relationships by relative asymmetric induction in the reaction of allylically substituted alkenes with electrophiles is of considerable importance in organic synthesis.<sup>1,2</sup> Although acyclic allylic alcohols have been widely studied,<sup>1</sup> and impressive selectivities recorded,<sup>1,3</sup> stereocontrol in related reactions of allylic amines and derivatives with electrophiles remains largely unexplored.<sup>4</sup> We had occasion to investigate this issue during our current synthetic efforts<sup>5,6</sup> directed at members of the pumiliotoxin A alkaloid class (1). For these studies we required enantio-



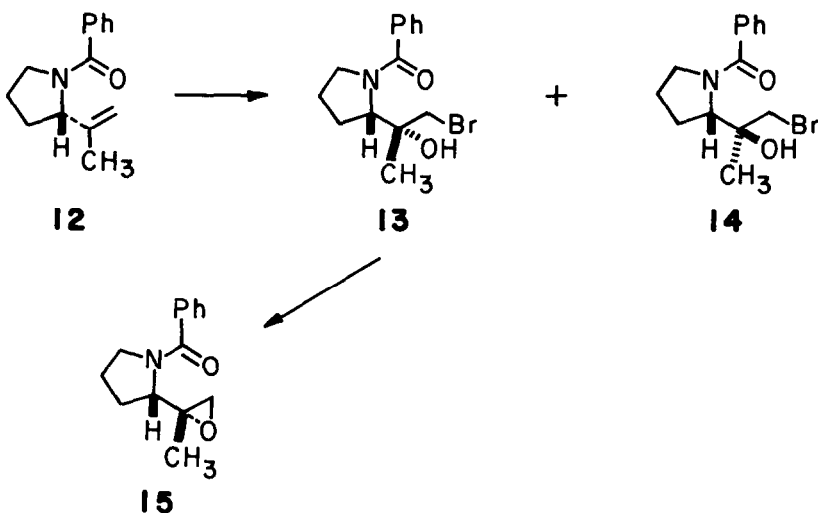
merically pure synthons 2, which should be available from L-proline. In this Letter we report that, although peracid epoxidations of N-acyl-2-isopropenylpyrrolidines (3, R'=COR) occur with little asymmetric induction,<sup>5</sup> halocyclization reactions<sup>7</sup> proceed with good selectivity and provide a stereocontrolled route for the synthesis of enantiomerically pure epoxides 2.

Asymmetric induction in the epoxidation of carbamate 4<sup>5</sup> with m-chloroperbenzoic acid was somewhat solvent dependent,<sup>8</sup> although at best (25°C in hexane) only a 2:1 ratio of epoxides 5<sup>5</sup> and 6<sup>5</sup> was realized. Similar low stereoselec-



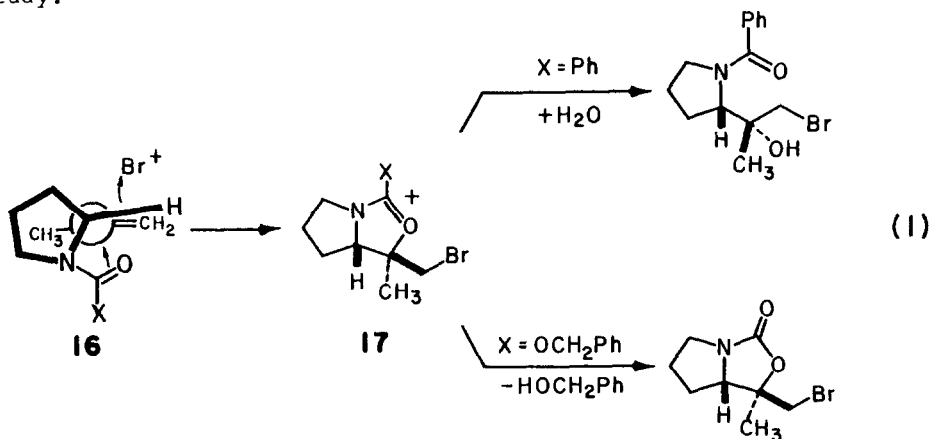
tivity was observed in the bis-hydroxylation of 4 with  $\text{OsO}_4$ . In marked contrast, 4 reacted with N-bromosuccinimide (NBS) in  $\text{DMSO-H}_2\text{O}$  (50:1,  $25^\circ\text{C}$ )<sup>9</sup> to give a single<sup>10</sup> bicyclic bromocarbamate 7<sup>11</sup> (mp  $75-76^\circ\text{C}$ ;  $^1\text{H NMR}$   $\delta$  1.54, s, Me, 3.52 ABq,  $\text{CH}_2\text{Br}$ ) which was isolated in 83% yield. That 7 had the same absolute configuration at the quaternary carbon as the desired synthons 2 was determined by conversion of 7 to deuterocarbamate 8<sup>11</sup> (mp  $56-58^\circ\text{C}$ , 82%) upon reaction with  $\text{Bu}_3\text{SnD}$  (PhH,  $50^\circ\text{C}$ ). The  $^1\text{H}$  decoupled  $^{13}\text{C}$  NMR spectrum of 8 showed a singlet for the  $\text{CH}_3$  group at 23.0 ppm, and a characteristic triplet for the  $\text{CH}_2\text{D}$  group at 28.6 ppm. The stereochemical assignment follows from our previous experience<sup>5</sup> that groups on the  $\alpha$ -face of this rigid bicyclic ring system experience upfield  $^{13}\text{C}$  NMR shifts. That the carbamate group of 4 was directly participating in the bromination step was established by treatment of a 1:1 mixture of 4 and alcohol 9 (a model for the bromohydrin of 4) with NBS in  $\text{DMSO-H}_2\text{O}$  to give bromocarbamate 7 and recovered alcohol 9 (carbamate 10 could not be detected by  $^1\text{H}$  NMR, GLC, or TLC analysis). Similar selectivity was observed when 4 was treated with iodine ( $25^\circ\text{C}$ ,  $\text{CH}_3\text{CN}$ ) to give a single<sup>10</sup> iodocarbamate 11<sup>11</sup> (mp  $83-84^\circ\text{C}$ ;  $^1\text{H NMR}$   $\delta$  1.47 s, Me, 3.44 ABq,  $\text{CH}_2\text{I}$ ) in 84% yield.<sup>12</sup>

Similar, although considerably reduced, selectivity was observed in bromohydrin formation from benzamide 12.<sup>13,14</sup> Thus, treatment of 12 with NBS in  $\text{DMSO-H}_2\text{O}$  ( $25^\circ\text{C}$ , 50:1) afforded a 7:3 mixture of bromohydrins 13 ( $^1\text{H NMR}$   $\delta$  1.31, s, Me) and 14 ( $^1\text{H NMR}$   $\delta$  1.25, s, Me) in 83% yield. The selectivity was increased to 9:1 when the reaction was conducted at  $-10^\circ\text{C}$  in  $\text{THF-H}_2\text{O}$  (50:1), and the major isomer 13<sup>11</sup> (mp  $104-104.5^\circ\text{C}$ ) could be isolated in 50% yield after chromatography



on silica gel. The stereochemistry of 13 was established by correlation<sup>15</sup> with 8. The desired epoxide 15<sup>11</sup> (an oil; <sup>1</sup>H NMR  $\delta$  1.44, s, Me, 2.80 ABq, CH<sub>2</sub>O, 4.59, t, J = 7.4, CHN) was obtained from 13 in 75% yield by reaction for 2h at room temperature with 0.3 M NaOH (5:2 THF-H<sub>2</sub>O) followed by rapid chromatography on silica gel.

The high stereoselectivity observed in the reaction of allylic carbamate 4 and allylic benzamide 12 with NBS and H<sub>2</sub>O can be rationalized as illustrated in eq 1.<sup>16,17</sup> Preferential reaction of conformer 16 is consistent with the "eclipsed alkene" model<sup>1,3,18</sup> of reactions of this type, as well as a recent computational study.<sup>2</sup>



**Acknowledgment:** Financial assistance from the National Institutes of Health (HL-25854) and the Camille and Henry Dreyfus Foundation (teacher-scholar award to L.E.O.) is gratefully acknowledged. NMR and mass spectra were determined with spectrometers purchased with the assistance of NSF instrumentation grants.

## References and Notes

1. For a recent review, see Bartlett, P.A. Tetrahedron **1980**, 36, 2-72.
2. For a recent theoretical analysis, see Caramella, P.; Rondan, N.G.; Paddon-Row, M.N.; Houk, K.N. J. Am. Chem. Soc. **1981**, 103, 2438-2440.
3. Cf. Sharpless, K.B.; Verhoeven, T.R. Aldrichimica Acta **1979**, 12, 63. Kishi, Y. Aldrichimica Acta **1980**, 13, 23-30.
4. (a) For a recent example in a cyclic system, see Georges, M.; Fraser-Reid, B. Tetrahedron Lett. **1981**, 22, 4635-4638; (b) For an independent investigation of the halocyclization of acyclic allylic carbamates, see Parker, K.A.; O'Fee, P.P. manuscript submitted for publication.
5. For the total synthesis of enantiomerically pure toxin 251 D (1, R=n-C<sub>3</sub>H<sub>7</sub>) see Overman, L.E.; Bell, K.L. J. Am. Chem. Soc. **1981**, 103, 1851-1853.
6. For current efforts directed at pumiliotoxin B (1, R=CH=C(Me)CH(OH)CH(OH)Me), see Overman, L.E.; McCready, R.J. Tetrahedron Lett. **1982**, 23, 2355-2358.
7. For a review of halolactonization, see Dowle, M.D.; Davies, D.I. Chem. Soc. Rev. **1979**, 171-197.
8. Ratios of 5:6 as determined by HPLC analysis of epoxidation reactions conducted at 25°C were: 52:48 (CH<sub>2</sub>Cl<sub>2</sub> or CH<sub>3</sub>CN), 58:42 (EtOAc), 59:41 (THF), 65:35 (hexane).
9. Cf. Dalton, D.R.; Dutta, V.P.; Jones, D.C. J. Am. Chem. Soc. **1968**, 90, 5498-5501.
10. The ratio of the two possible diastereomers was greater than 20:1 (<sup>1</sup>H NMR).
11. All new compounds showed IR, 250 MHz <sup>1</sup>H NMR (CDCl<sub>3</sub>), 63 MHz <sup>13</sup>C NMR (CDCl<sub>3</sub>), and mass spectra which were fully in accord with their assigned structures.
12. This reaction was first conducted by Dr. Kenward Vaughan.
13. Mp, <sup>10</sup>1-102°C; prepared from N-benzoyl L-proline in a fashion similar to 4.
14. Direct epoxidation of 12 with m-chloroperbenzoic acid was nonstereoselective.
15. Accomplished in 25% overall yield by the following sequence: (a) Bu<sub>3</sub>SnD, PhH, 50°C; (b) DIBAL-H, 25°C, PhMe; H<sub>2</sub>, Pd-C; (c) PhCH<sub>2</sub>OCOC<sub>2</sub>H<sub>5</sub>, Et<sub>3</sub>N, 25°C; BuLi, THF, 25°C.
16. The lower stereoselectivity observed with 12 may reflect competing bromohydrin formation by a nonstereoselective reaction which does not involve carbonyl participation.
17. The somewhat related formation of halolactones from the reaction of 3-butenylcarboxamides with I<sub>2</sub> and H<sub>2</sub>O has been reported: Takano, S.; Hirama, M.; Ogasawara, K. J. Org. Chem. **1980**, 45, 3729-3730.
18. For a recent example, see Collum, D.B.; McDonald, III, J.H.; Still, W.C. J. Am. Chem. Soc. **1980**, 102, 2118-2120.

(Received in USA 13 August 1982)